

Removal Efficiencies of a Polymer Enhanced Dewatering System

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Senior Associate



Site terrain in Western Canada

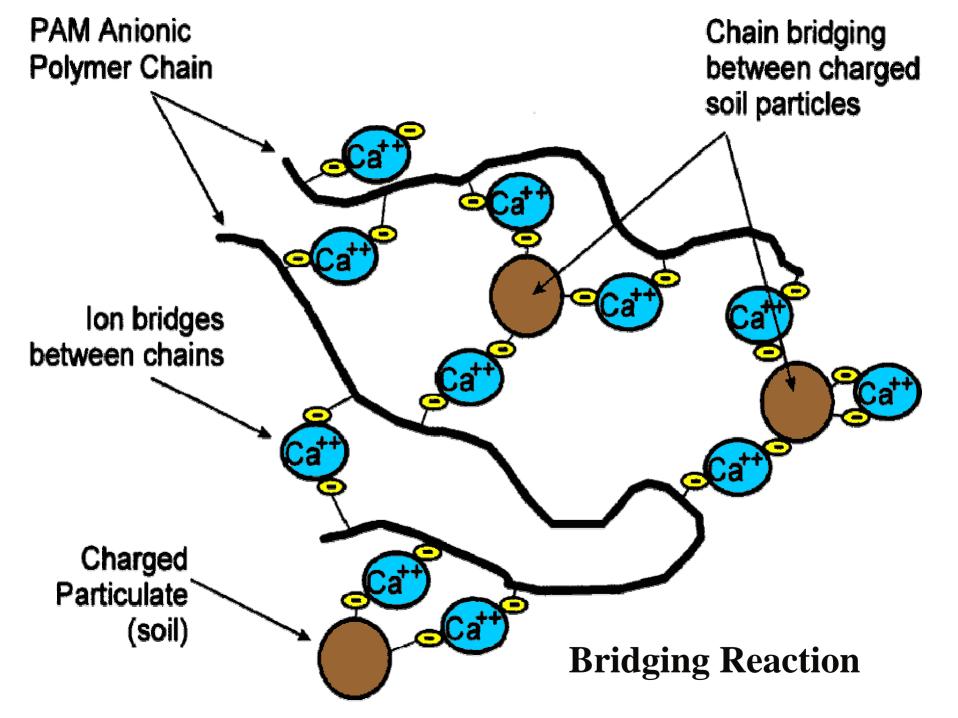




Pond showing elevated colloidal clay content

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Blinding of polymer log due to high CaCo3 concentration

2005/10/06

Correct polymer type showing log dissolving

> **Incorrect polymer type showing log not dissolving**

Mixing chamber installation at -15 Celsius

Mixing chamber installation at requires only basic Form Carpentry techniques

Mixing chamber must be installed level

Back fill can wait until spring

Installation was based on gravity flow



Example photo showing mixer + plunge pool and Baffle grid for particle collection when ponds are not available

Example photo showing baffle grid in place of a pond



Metal Removal Data

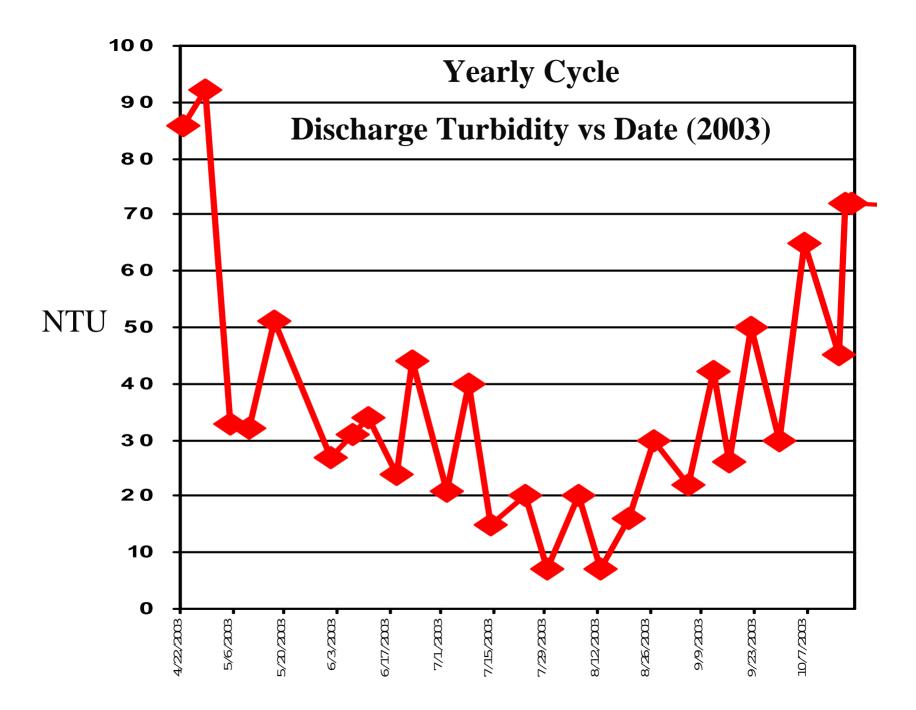
Floc Logs®



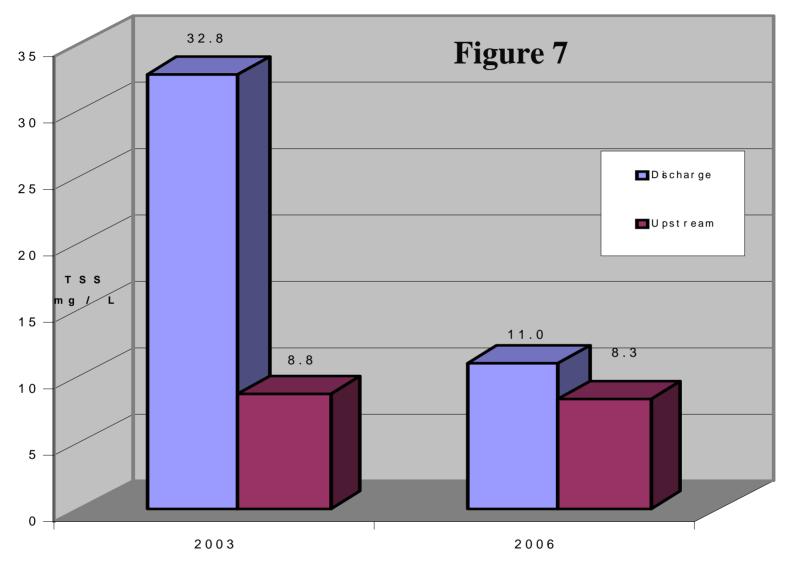


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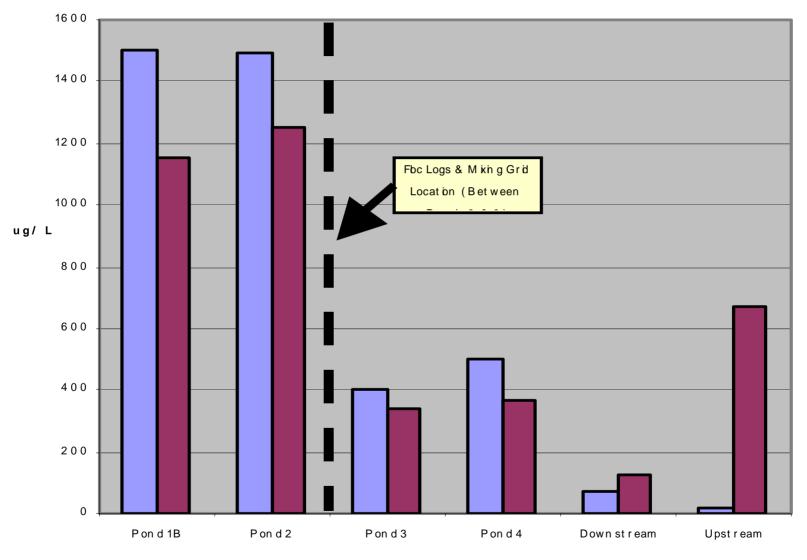
Discharge - Average Annual TSS (mg/L)





Average Annual TSS before and after Polacrylamide

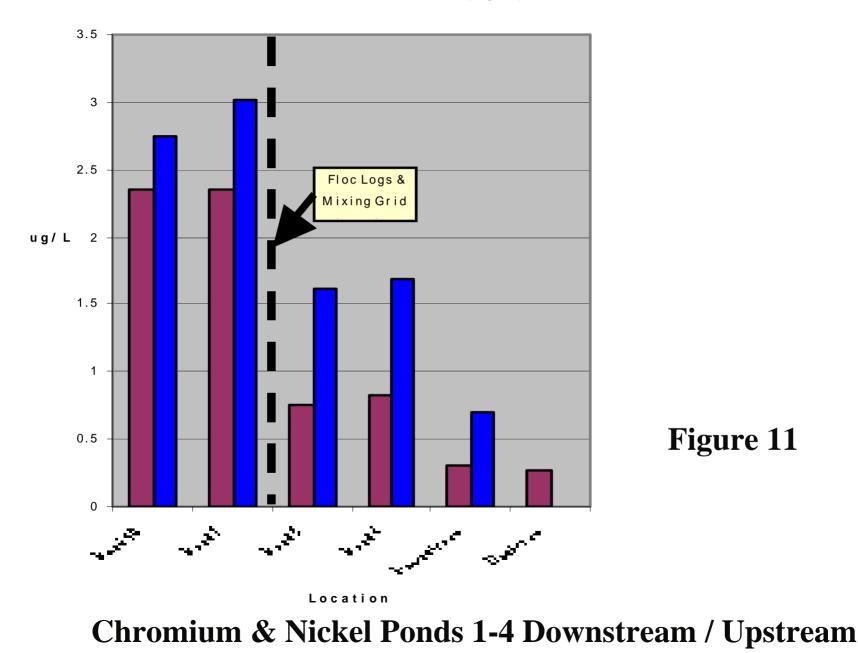
Wash Plant Pond Sample Results Recoverable Metals (ug/L) - Oct 12, 2006 Figure 9



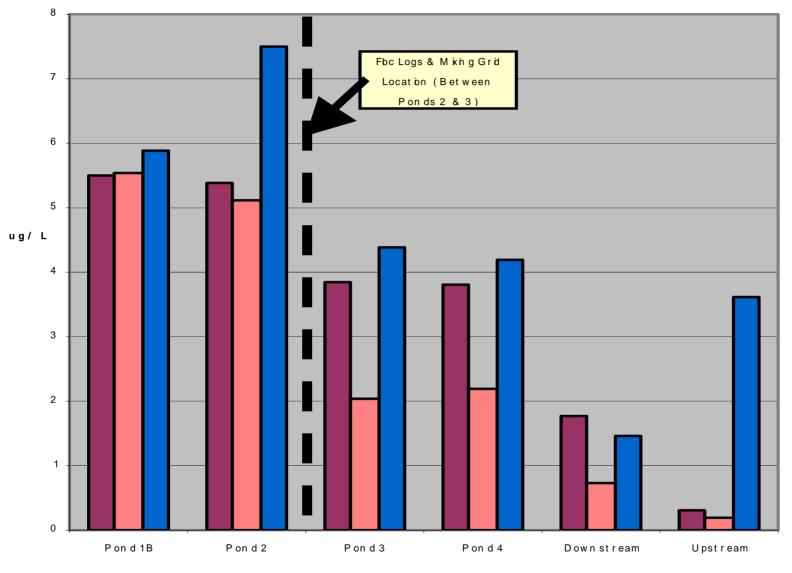
Location

Aluminum & Iron Ponds 1-4 Downstream / Upstream

Wash Plant Pond Sample Results Recoverable Metals (ug/L) - Oct 12, 2006



Wash Plant Pond Sample Results Recoverable Metals (ug/L) - Oct 12, 2006 Figure 10



Location

Copper, Vanadium & Zinc Ponds 1-4 Downstream / Upstream

Wash Plant Pond Sample Results Recoverable Metals (ug/ L), Oct 12, 2006

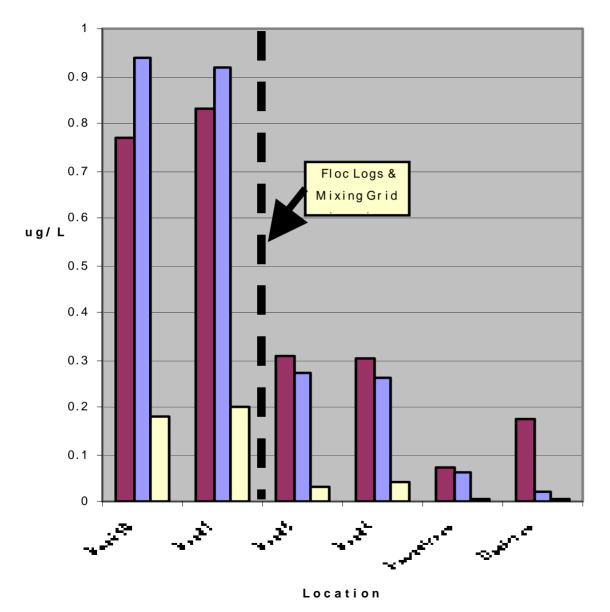
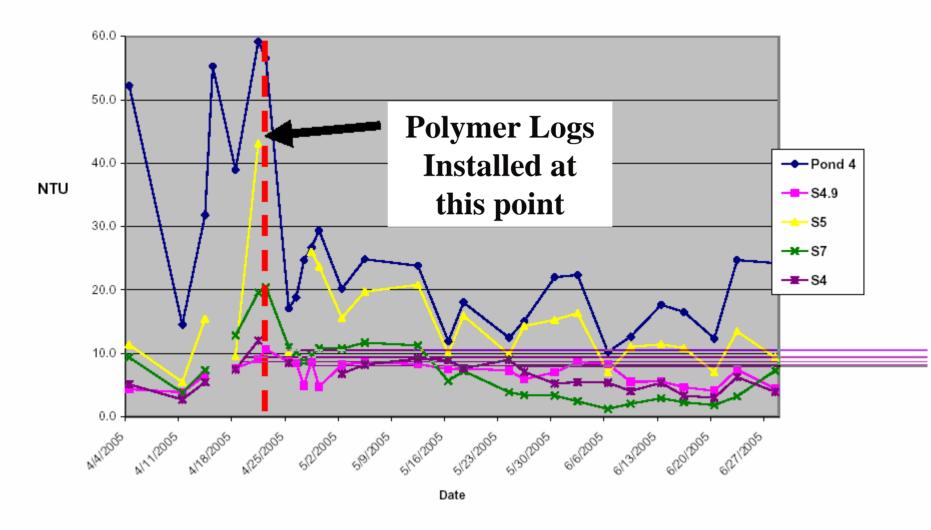
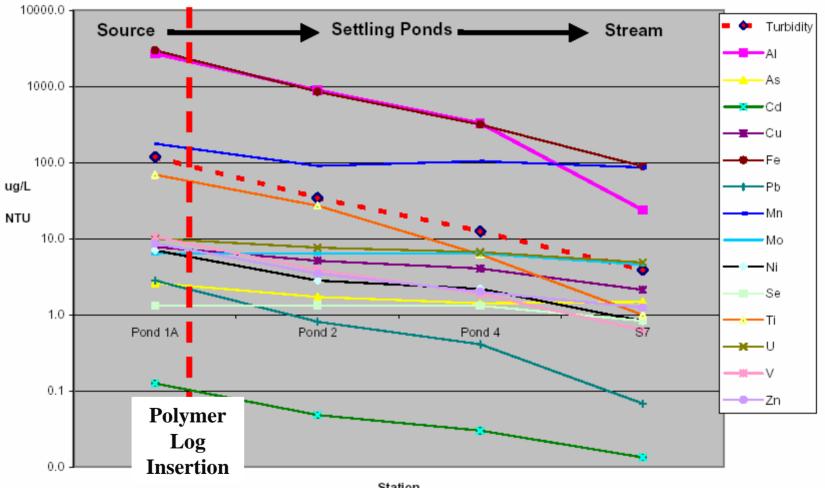


Figure 12

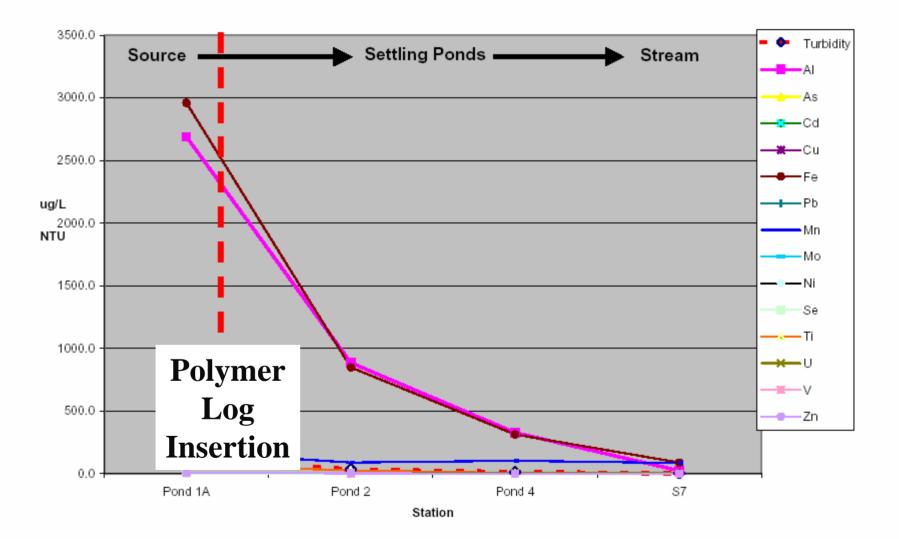
Cobalt, Lead & Thorium Ponds 1-4 Downstream / Upstream

Wash Plant - Turbidity 2005





Station



Element	% Change in metals Pond 1 - 2	% Change in metals Pond 2 - 3 (Polymers Inserted)	% Change in metals Pond 3 - 4
Th	12%	-86%	39%
Ti	-1%	-74%	67%
Al	-1%	-73%	25%
Fe	9%	-73%	8%
Be	-4%	-71%	-14%
Pb	-3%	-70%	-3%
Cr	0%	-68%	9%
Bi	-14%	-65%	-15%
Со	8%	-63%	-2%
V	-7%	-61%	9%
Ag	-8%	-58%	23%
Ni	9%	-46%	4%
Zn	27%	-41%	-5%
Cd	-8%	-37%	-9%
Cu	-2%	-28%	-2%
As	10%	-25%	-2%
Mn	-4%	-22%	5%
T1	-3%	-21%	-1%
Sn	-3%	-18%	-5%
Sb	-11%	-12%	-1%

Ponds 1&2 show metal increases likely due to pond soil contamination.

Polymer log installations feeding to ponds 3&4 show an overall decrease of metals.

Represents Increase in Metals Values

Figure 5: Percentage Reduction/Increase in metals in settling ponds, sample date Oct 6, 2006

Rules of Polyacrylamide Use

- Variations of polymer length, electrostatic charge, type of polyacrylamide and additives all effect the performance of the polymer
- Each soil chemistry is unique and requires adjustment to the polymer mix to assure best performance
- Greater application rates <u>do not</u> result in better performance
- Correct BMP combinations with polymer use result in best performance
- Each polymer mix requires EPA certified toxicity reports to assure absence of aquatic toxicity
- Performance testing before use or application must show 95% or better attachment to the soil to assure correct polymer





Rules of Polymer Use

1) Polymer must be non-toxic to aquatic organisms having EPA certified toxicity reports (whole product WET tests using ASTM guidelines)

2) Each site application must demonstrate 95% or better NTU reduction test reports

3) Each polymer can be unique for each application. One polymer does not work on all soils











Questions you should be asking

Do you have the acute and chronic ASTM-EPA aquatic toxicity reports for the intended polymer?

Have you performed site specific jar testing, or lithology testing achieving 95% or better results to show that the polymer will work on the particular site where application is intended?

If so, please proceed and help keep our

environment clean.





Senior Associate





Summary

BMPs must be used in combinations

One BMP will usually not produce compliance results

Jute matting or equal should be used with all types of polymer applications



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REFERENCE WEB SITES

http://kimberly.ars.usda.gov/pampage.shtml

http://kimberly.ars.usda.gov/Pamprim.shtml

www.siltstop.com

www.stormwater.ucf.edu

Other information available on request





